Madres Agric J. 71 (5) 302-310 May 1984

# GENETIC PANS OF THREE SOILS OF THE SOUTHERN COASTAL PLAINS OF THE UNITED STATES

K. G. PRASAD\* and H, F. PERKINS

An investigation was undertaken to determine morphological, physical, chemical, and mineralogical properties of pans having fragic characteristics in horizons of selected Southern Coastal Plain Soils, Soil series involved were Carnegie (Plinthic Paleudults), Cowarts (Typic Hapludults) and Irvington (Plinthic Fragiudults). Pans were strongly indurated at the top and hardness gradually decreased with depth. Pans were firm when moist and very hard when dry and ranged from 15 to 45 cm in thickness. There was no evidence that particle size distribution was related to pan formation. Organic matter Content, pH, and available P of the epipedon were higher than subsurface horizons although magnitude varied from soil to soil. Cation exchange capacity was fairly consistent in a given horizon for each pedon and was related to organic matter and clay content. Exchangeable Mg content of pans was higher than exchangeable Ca. Pans did not influence the distribution of exchangeable Na, K, Zn, Fe, and Mn. Horizons having fragic characteristics were not related to gross soil mineralogy or clay mineralogy. Close packing of particles and possible surface tension effects due to the structure of the water molecule apparently influenced the bonding which, resulted in the brittleness in the pan formations. (Key words: Genetic pans, Gragipans, pans, Sub surface horizon).

Fragipans are considered genetic soil horizons (soil survey staff 1973). They are subsurface features which may vary widely in thickness, Excepton nearly level topography, fragipans usually occur 25 to 100 cm below the surface of the soil. On very level areas, the top of the pan may be within 35 cm of the surface (Winters and simonson 1951). Usually pans nearer the surface are more distinct than those deeper in the pedon. The presence of a pan in a soil has an unfavorable influence on plants by restricting root growth, either through mechanical impedance or by creating a high water table (Prasad and Perkins 1978) during parts of the year. The critical depth of a Pan depends on the type of plants. For many shallow rooted plants, the influence diminishes rapidly when the pan occurs at depths greater than 50 Cm (Grossman and Carlisle 1969).

Fragipans have been reported to have low pH values and correspondingly low base saturation (Jha and Cline 1963. Total quantities of Ca, Mg, K, and P in the pans are similar to those in other subsurface horizons (Winters and Simonson 1951). Fragipan development seems unrelated to the mineralogy of the sand and silt fraction (Carlisle et al 1957). The clay mineralogy of fragipans is similar to the clay mineralogy of horizons in associated soils without pans (Grossman and Carlisle 1969).

Department of Agronomy, University of Georgia, Athors Georgia, U.S.A.

<sup>\*</sup> Present Address: Forest Soil Cum Vegetation Survey, Southern Region Coimbatore-641002.

do not lie in any single feature, but in a group of attributes which together make it a pan. The purpose of this paper was to study selected characteristics of genetically formed pans frequently referred to as fragipans and associated horizons in three Southern Coastal plain soils.

### MATERIALS AND METHODS

Study areas are located in the Southern Coastal Plain soil province of Georgia. Three pedons each of the Carnegie (Plinthic Paleudults Cowarts), (Typic Hapludults) and Irvington (Plinthic Fragiudults) series were described and samples from each genetic horizon ,were collected (Soil Survey Staff 1972) Sand fractions were separated by dry sieving (Soil Survey Staff 1972) and silt and clay particles were separated by centrifugation (Jackson 1956). Soil pH was determined in deionized water and in 0.01 N CaCl. in a 1:2 soil to solution ratio Organic matter (wet oxidation); Available P (extracted by HCL-HISO, mixture and determined by the vanadomolydophosphoric yellow color method; Cation exchange capacity (CEC) (1N NH, OACpH 7), and exchangeable cations were determined by atomic absoption spectrophotometry in the NH1 OAC leachate as described by Jackson (1958). Exchan geable acidity and percent saturation calculated based on CEC and the sum of exchangeable cations Mineralogy of the clay and silt fractions were determined by DTA and X-ray diffraction. A self-recording instrument with 3 mg samples, r-Al+0: standard, and a temperature increase C/minute was used in the DTA. For

crystalline mineral species identification, samples were analysed on a X ray diffraction unit employing CuK oc radiation. Heavy minerals were separated in 0.1 0.05 mm fraction using bromoform (density 2 8 g/cc at 20 C) (Matelski 1951), mounted on a gelatin coated slide and identified with a polarizing microscope (Jackson 1956).

## RESULTS AND DISCUSSION

Important morphological features of representative pedons studied are summarized in Table 1, similarity in structure existed in the pan layers. The pans were strongly indurated and brittle at the top which gradually decreased with depth. Open cracks in the pans were not evident. Pans were divided into large prismlike structural units by approximately vertical zones of light colored streaks On a horizontal section through the panlight colored streaks formed a polygonal pattern. The pans were firm when moist and very hard to extremely hard when dry. Thickness of the pans ranged from 15 to 45 cm. Morphologically, the pans are more strongly expressed in the Irvington than in the Carnegie and Cowarts soils. Morphological features of horizons containging pans conform to the diagnostic features of fragipans described (Jha and Cline 1963, and Lozet and Herbilon 1971).

Data in Table 2 suggest significant eluviation of clay from the epipedon to the subsurface horizons percentages of clay free sand among horizons indicate that the parent material for the existing soil was uniform throughout. In general fragipans tend to be medium textured

(Grossman and Carlisle 1969), however textural class usually depends on parent material and degree of weathering (Jha and Cline 1963). It has been reported that fragipans have no influence on particle size distribution in soil pedon (Grossman and Carlisle 1969).

Values for PH, organic matter, and available p of the surface layers were higher than for subsurface hoizons although the magnitude varied from soil) Table 3). Uniformly lower pH of soil measured in 0 01 N CaCl solution as compared to H.O signified the dominance of negatively charged exchange sites throughout the pedon. Acid extractable p appeared to be related to organic matter, being relatively high in the surface and consistently low in the remainder of the pedon. (Jha and Cline 1963, and Lozet and Herbilon 1971).

CEC values were fairly consistent in a given horizon for each 'pedon' (Table 3). Values in the A horizons of most soils reflected an accumulation of organic matter. High clay content in the Bt horizons accounts for the relatively high CEC values in these horizons irrespective of the presence or absence of fragipans (Table 2 and 3). Exchangeable Mg content of pans was higher than exchangeable Ca In general, exchangeable Ca was inversely related to exchangeable Mg in the pedon with depth. Hutcheson et al (1959) suggested that high extractable Mg may make clay more susceptible to movement and rearrangement, thereby fostering horizon rigidity. The pan didnot influence the distribution of other cations. Data indicated little chemical

change from the top of the pan down ward and chemical change above the pan appeared to be associated primarily with organic matter and clay content. Jha and Cline (1963) and Lozet and Herbilon (1971) also reported similar results for other soils containing pan formations.

Distribution of heavy/minerals with depth is shown in Fig. 1. There was a marked increase in heavy mineral content in the argillic horizons in two of the three pedons of the Carnegie soils. Most of these heavy minerals were hematite. The concentration of hematite was related to the presence of plinthite. A similar trend, but with less magnitude, was also observed in cowarts and Irvington soils. Hematite in the silt fraction of plinthite from Carnegie, Cowarts and Irvington soils has also been reported. elsewhere (Wood and Perkins 1976). Nikiforoff et al 1948) theorized that significance of weight percentage of heavyminerals in hard pan soils was obscured by the large amounts of secondary iron minerals in the B- horizon and the pan. The light separates were primarily quartz but contained trace amounts of the feldspars. Daniels et al (1966) found no relationship between fragipan expression and the percentage of feldspar in the very fine sand fraction for certain soils of North Carolina, Hutcheson et al 1959, and Nikiforoff et al 1948 reported that high proportions of quartz do not appear to directly contribute to the prevalence of pans in Coastal plain soils.

In subsurface horizons, kaolinite was the dominant silicate clay mineral of the clay fraction (Table 4). There was a

slight increase in kaolinite content with depth in all pedons. Small amounts of goethite, gibbsite, and intergrade micaceous minerals were also identified. In the fragipon soils of the northeastern United States. 2:1 lattice clay other than montmorillonite predominates, with mica a quantitatively important component (Jha and Cline 1963). Illite, chlorite, and interestratified minerals with some interlavered montmorillonite in the clay fraction of several soils with fragipans were reported from Michigan and northeastern Wisconsin (Grossman and Carlisle 1969). Soils with pans in the middle and lower Mississippi Valley developed in loess contain appreciable amounts of montmorillonitei (Hutcheson et al 1959) In Maryland, significant amounts of kaolinite were found in the pan soils (Nikiforoff et al 1948).

It has been proposed that silicate clay is the principal bonding agent in fragipans (Hutcheson et al 1959, Jha and Cline 1963 and Grossman and Carlisle 1969). Close packing of sand and silt as observed in the current study. is thought to contribute to the effectiveness of clay as a bonding agent through interlocking of sand silt and (Nikiforff et al 1948), or like brick and mortar Structure (Jha and Cline 1963) or as clay bridges (Lozet and Herbilon 1971) However, there is no evidence that the clay in fragipans resists disaggregation (Grossman and Carlisle 1969). Hydrated oxides of Fe or Al and amorphous Sioxides have been suggested as bending agents (Grossman and Carlisle 1969, and vinters and Simonson 1951). Surface ension effects associated with water ilms have also been proposed to lend

rigidity to soil material (Grossman and Carlisle 1969). Data published by the authors does not indicate that Si, Fe, and Al, or silicate clay minerals are responsible for bonding (Prasad and Perkins 1983). However, close packing of sand and silt particles and possible surface tension effects due to structured water molecules cannot be ruled out. Infrared analysis indicated the presence of hydrogen-bonded hydroxyls which disapp eared after. heating to 300 C. Brittleness restored by rehydration under moderate relative humidity. These findings confirm the hypotheses of surface Tension effects associated with water films.

Fragipans have been reported to occur on old geomorphic surfaces (Daniels et al 1966; and Nikiforoff et al 1948). Inheritance of properties in fragipan development on these surfaces have been suggested however, current studies indcate that pans occurring in Southern Coastal Plain soils do not inherif fragic characteristics from parent material Incremental development of fragipans has been proposed dy Daniels et al (1966). but, the incremental development leads to the question of time required to form a fragipan (Grossman and Cartisle 1969) Processes common to a periglacia environment have been implicated in the development of fragipans (Lozet and Herbilon 1971), but arguments against periglacial origin have also been presented (Jha add Carlisle 1969).

In view of the fact that a specific, cementing agent was not detected, and a high bulk density and poor pore size distribution exists (Prasad and Perkins 1978), suggest that the fragic properties

# PRASAD and PERKINS

Table 1: Morphological descriptions of representative pedons from Carnegie, Cowarts, and Irvington soils.

Horizon	Depth	Munsell color Texture Structure Consistence Boundary Comments
		- Carnegles Series
Apo Bic Bix	0-15 15-48 48-91	10YR 4/3 s 1 w. fi & med, gr V Fr aw 10% Fe cond 10YR 5/8 sc1 w, med, sbK Fr gw 5% Fe cond 2.5YR 6/6, cm,med,pr sc1 mod, med, pt, 2.5YR 3/6 to sbk 10% plinthits 60% brittlenese
B11	91-112	10YR 6/8, cm,med,pr sc1 W to mod med, Fr gw. 3% plinthite 10YR 4/6 and sbk 2. 5YR 4/8, cm,med di, 2. 5YR 8/2
B12	112-157	10YR 5/8, m,co.pr. sc1 w, med, sbk Fr gw 3% plinthite 2.5 8/2 and 2.5YR 4/8 cm, med, fa 10YR 6/6
	9 - 4	Cowarts Series
Ap	0-15	2.5YR 6/2 1s w, fi, gr V Fr cs few quertz gravel
ВА	15 36	10YR 5/4 s1 w, fi, gr Fr ci few quartz gravel
Btx	36-56	10YR 5/6, sc1 mod.co.pt.to Br cw 15% plinthite cm, med to mod, med, abk 60% brittleness 4/6 and cm, med,ta,10YR 6/6
Bt1	56-71	10YR 5/8, fe, fi, co sc1 mob, co.pt to. Fi cw 3½ plinthite fa, 2.5/YR 4/8, mod, co. abk and cm med,d, 5Y 7/2
Bt2	71-152	10YR 5/6, 5YR co sc1 w,co,abk FI 3% plinthit 6/4 and 5Y 7/1 Irvington Series
Ap Btc1 Btc2 Bix	0-33 33-41 41-76 76-155	10YR 4/2 1s w, med ,gr V Fr cw 4% Fe cond 10YR 6/6 st w, med, sbk Fr cs 15% Fe cond 10YR 6/6 sc1 mod, meb, sbk Fi gw 40% Fe cond 25Y 7/4, sc1 mod, med, abk Fi, Br — 20% Fe cond 15% dlinthite 10YR 6/8, cm, med, d, 5R 5/8
		The state of the s

Mottles: fe = few, cm = common, m, = many, fi = fine, med, = medium, co = coarse, fa = faint di = distinct, pr = prominent

Texture: s = sand/sandy, c = clay/clayey, l = loam/loamy, co = coarse

Structure: W = weak, mod = moderate, fi - fine, med = medium, co = coarse, gr = granular sbk = s b-angular blocky, abk = angular blocky, pt = platy, pr = prismatic

Consistence: Br = brittle, Fi = firm, Fr = friable, V Fr = very friable

Boundary: e = abrupt, c = cear, g = gradual, i = irregular, s = smooth, w = wavy

Table 2. Particle size distribution in representative pedons from Carnegie, Cowarts, and Irvington Soils.

45.0			Particle size	ze (mm)		7	
Horizn 5.0		0.5—	0.25-	0.1—	Total	Total	. Total
10	0.5	0.25	0.10	0.05	Sand	Silt	Clay
(				<del></del> %			
			Carnegie	41.7	80 er		7
Apc 8	18 13	24 17	22	- 8	80 59 63 57	7	13
Bfc 4.	13	17	18 18 14 12	7	59	11 9 14 13	30
Btx · 4	14	20	18	. 7	63	9 .	28
Bt1 4	14	19	14 -	6 5	57	14	29
Bt2 5	15	19	12	5	56	13	31
	-		Cowarts	Series			
Ap 3	7	19	34	14	77 61 53 59	-11	12
Ap 3 BA 3	' 7	19 16	26 .	14 9 7	61	11	- 28
Brx 3	9 14 18	17	34 26 17 13 9	7	53	11	28 36 30
B1 10	14	17 13	13	5	59	11	30
Bt2 19	18	13	9	4 -	63	9	28
			Irvington	Series	71		
Ap 1	4	2 16 16 17	41 34 34	15	63	31 9	6
Btc1 1	4 3 3 3	. 16	34	12	66 65	9	6 25
Btc2 1	3	16	34	11 9	65	10	25 24
Btx .1	3	17	20	9	50	26	24

Table 3. Selected chemical properties of selected pedons from Carnegie, Cowarts, and Irvington soils.

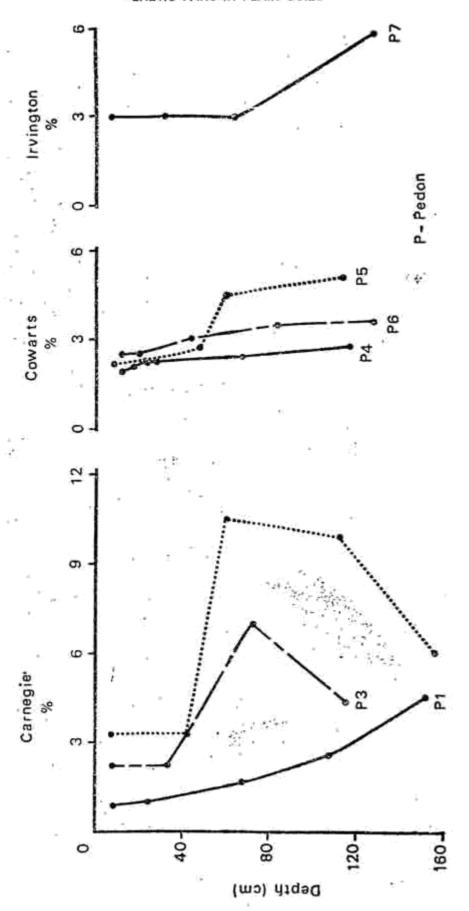
4 10	PH (1:2)		Org.		Ext.		Exchangeable					Base		
Horizon	H <sub>1</sub> O	CaC1		matter P	CEC	Ca	Mg	Na	K	Zn	Fe	Mn	acidity	saturation
		(%	/ <sub>0</sub> ) (με	g/g) (		meq/	100g_	-	_) (.	μ	g/g	) (	(meq/10	0g) (%)
					Carne	egie	Serl	es						
Apc better Btx Bt1 Bt2	5.2 5.1 5.1 4.9 4.7	4.1 4.1 4.1 4.0 3.9	3.9 0.2 0.2 0.2 0.2	8.5 1 8 2.1 1 8 0.7	4 4 5.3 6.3 4.0 3 5	1.8 1.3 1.0 0.9 0.6	0.8 1.1 1.5 1.0 0.5	0.8 0.3 0.3 0.4 0.4	0.	3 1 0 2 1.8 ! 1.3	3.8	0.3	2.3 3.3 1.5	86 57 48 38 51
.—. 1	***************************************	+			Co	warts	S	eries			- 40			
Ap BA Btx Bt1 Bt2	4.8 4.8 4.9 4.9 4.6	4.0 3.9 3.9 3.8 3.8	2.0 0.4 0.3	21.1 9.5 0.7 0.7 0.7	5.0 4.8 6,0 5.8 4.3	1.4 0.9 0.6 0.2 0.4	1.1 1.0 1.4 0.9 0.8	0.3 0.4 0.5 0.5	0.5 0.2 0.2 0.2 0.2	1.5 0.5 0.3 0.4 4.7	2,8 3.8 3.4 2.0 3,2	3.0 1.4 0.2 0.2 0.4	2.4	66 50 43 31 44
					trvin	gton	Se	ries						
Ap Btc1 Btc2 Btx	4.8 5.1 4.9 4.8	4.0 4.3 4.1 4.0	2.3 0.5 0.4 0.3	12.3 4.2 1.8 0.7	4.8 4.8 5.0 4.5	1.2 1.0 0.9 0.4	0.8 0.9 1.0 0.5	0.3 0.4 0.4 0.2	0.5 0.2 0.1 0.1	2.3 0.8 0.8 1.1	2.8 5.6 5.1 5.2	6 5 0.9 0 5 0.4	2.0 2.3 2.6 3.3	58 52 48 27

Table 4 Silt and clay minerlogy of representative pedons from Carnegie, Cowarts:

and Irvington soils.

Horizon		Particle Size (mm)									
	0.5-0.02	0.02-0.005	0.005-0.002	0.002-0.0002							
			Carnegie Series								
Apc	Q <sub>1</sub> *	Q <sub>1</sub> K <sub>4</sub> Gi <sub>4</sub>	K <sub>2</sub> Q <sub>2</sub> Gi <sub>2</sub>	K₁Gi₀Q, K₁Gi₀							
Btc	Q,	Q,K,Gi,Go,Mi,	K₃Q₃Gi₄Go₁Mi₁	KıGi.Go.Mi.Q. KıGi.Go.Mi.							
Btx	O1K1	Q <sub>1</sub> K <sub>3</sub> Gi <sub>3</sub> Go <sub>4</sub> Mi <sub>4</sub>	K <sub>3</sub> Q <sub>8</sub> Gi <sub>3</sub> Go <sub>3</sub> MI <sub>4</sub>	KıGi GosMisQ. KıGosGisMis							
Bt1	Q <sub>1</sub> K <sub>3</sub>	Q,K,Go,Gi,Mi,	K,Q,Go,Gi,Mi,	KıGo,Gi,Mi,Q, KıGo,Gi,Mi,							
B12	Q <sub>1</sub> K <sub>1</sub> Go <sub>4</sub> Gi <sub>4</sub>	$Q_2K_2Go_0Gi_4Mi_4$	K,Q,Go,Mi,Gi,	KıGoses MîrOx KıGosGisMir							
		4	Cowarts Series								
P	סיגי	Q <sub>1</sub> K <sub>1</sub> Mi <sub>4</sub>	K <sub>2</sub> O <sub>2</sub> Go <sub>2</sub> Gi <sub>4</sub> Mi <sub>4</sub>	KıQıGorGirMir KıGosGirMir							
BA	Q <sub>1</sub> K <sub>3</sub> Go <sub>4</sub> Mi <sub>4</sub>	Q <sub>I</sub> K <sub>1</sub> Go <sub>2</sub> Mi <sub>4</sub>	K <sub>2</sub> Q <sub>3</sub> Go <sub>8</sub> Gi <sub>3</sub> Mi <sub>4</sub>	K <sub>1</sub> Go <sub>3</sub> Gi <sub>3</sub> Mi <sub>4</sub> Q <sub>4</sub> K <sub>1</sub> Go <sub>3</sub> Gi <sub>2</sub> Mi <sub>4</sub>							
Bix	$O_1K_3Go_4GI_4M$	i, Q,K,Go,Gi,Mi,	KıGoaGiaMiaQa	KıGoıGi,MiıQı KıGoıGi,Mi							
311	Q <sub>1</sub> K <sub>8</sub> Go <sub>4</sub> Gi <sub>4</sub> M	i₄ Q₁K₂Go₄Gi₄	K,Q,Go,Gi,Mi,	KIGOIGILMILOI KIGOIGILMIL							
112	Q1K1Go4	Q1K3G04	K1Q1Go1Mi+	KıGoaQıMi, KıGoı							
			Irvington Series	1 2							
νP.	Q <sub>1</sub>	Q1K4Mi*	Q <sub>2</sub> K <sub>3</sub> Mi <sub>3</sub>	K₂Q₁Go₃Mis K₁Q₁Go₃Ma							
tcl	Q:Mie	Q <sub>1</sub> K <sub>3</sub> Go <sub>4</sub> Gi <sub>4</sub> Mi <sub>4</sub>	K,Q,Mi,Go,Gi,	KıMisQıGogGir KıMisGogGir							
tç2	O1K4	Q <sub>1</sub> K <sub>2</sub> Mi <sub>4</sub> Go,	K <sub>2</sub> Q <sub>1</sub> MiaGo+	K <sub>1</sub> Mi3Go3Q <sub>4</sub> K <sub>1</sub> Go3Mi <sub>4</sub>							
tx	Q₁K₃Go₃Mi₄	Q <sub>1</sub> K <sub>3</sub> G <sub>0</sub> ,Gi,Mi,	KaQaGoaGi,Mi	KıGosGi,Mi,Qı KıGo,Gi,Mi,							

<sup>\*</sup> K = kaclinite, Gi = Gibbsito, Go = Goethite, Mi = Intergrade mica, Q = Quartz 1 => 50%, 2 = 50 to 30%, 3 = 30 to 10%, 4 = <10%



of horizons in southern Coastal plain soil are primarily a result of compression and orientation of soil particles.

#### REFERENCES

- DANIELS, R. B., W. D. NETTLETON.
  R.J. McCRACKEN, and E. E. 1966,
  Morphology of soils with fragipan in parta
  of Wilson Country, North Carolina, Soil Sci.
  Soc. Amer. Proc. 30: 376-380.
- GROSSMAN, R,B and F.J. CARLISLE 1969. Fragipan soils of the eastern United States Advance Agron. 21: 237-279.
- HUTCHESON, T.B., Jr., R.B. LEWIS, and W.A. SEAY. 1959. Chemical and mineralogical properties of certain Memphis catena soils of western Kentucky. Soil Sci. Soc. Amer. Proc. 23: 474-478.
- JACKSON, M.L. 1958. Soil chemical analysis Prentice Hall, Inc., N.J.
- JACKSON, M.L. 1956, Soil Chemical analysis.

  Advanced course. Publ. By the author

  Madison, Wisc.
- JHA, P.P., and M.G. CLINE. 1963. Morehology and genesis of a Sol Brun Acide with fragipan in uniform silty material. Soil Sci. Soc. Amer. Proc. 27: 339-344.
  - LOZET, J.M., and A.J. HERBILON. 1971
    Fragipan soils of Condroz (Belgium)
    Mineralogical, chemical, and physical
    aspects in relation with their genesis
    Geoderma 5: 325-343.

- MATELSKI, R.P. 1951. Separation of minerals by subdividing solidified bromoform aftercentrifugation. Soil Sci. 71: 269-272.
- R.P. HUMBERT. 1948. The hardpan in certain soils of the Coastal Plain. Soil Sci. 65: 135-153.
- PRASAD, K.G. and H.F. PERKINS. 1978. In situ hydraulic conductivity of a fraginar soil in the Southern Coastal plains. Soil Sci. 126: 263-268.

+ 10.

- PRASAD, K. G, and H.F. PERKINS. 1983
  Occurence and Characteristics of fraginan
  like formations in selected coastal plain
  soils. J. Ind. Soc. Soil Sci. 31: (In press).
- SOIL SURVEY STAFF. 1973. Soil Taxonomy Agric. Hand-book No. 436 Printing Office, Washington D.C.
- SOIL SURVEY STAFF. 1972. Soil survey laboratory methods and procedures for collecting soil samples. Soil Survey Investigation Reports No. 1. Govt. printing Office Washington, D.C.
- WINTERS, E., and R.W. SIMONSON. 1951. The subsoil. Advance. Agron. 3: 1-92.
- WOOD, B.W. and H.F. PERKINS. 1976
  Plinthite character-ization in selected
  Southern Coastal Plain soils. Soil Sci. Soc.
  Am.J. 40: 143-146.